# Efficient conversion of $2^{\prime}$-hydroxychalcones into flavanones and flavanols in a water suspension medium 

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Conversion of $2^{\prime}$-hydroxychalcones into flavanones and flavanols was found to proceed very efficiently in a water suspension medium.

## Introduction

Flavanones 2 are important naturally occurring pharmacological compounds and are valuable precursors for the synthesis of flavanoids. ${ }^{1}$ Preparation of flavanones $\mathbf{2}$ has been carried out by intramolecular cyclization of 2-hydroxychalcone 1 under various conditions using acids, ${ }^{2}$ bases, ${ }^{3}$ thermolysis, ${ }^{4}$ electrolysis ${ }^{5}$ and photolysis. ${ }^{6}$ However, the yields of these reactions are often moderate ( $20-90 \%$ yield) and the reaction usually gives a mixture of $\mathbf{1}$ and $\mathbf{2}$, the separation of which requires a lot of organic solvent such as benzene. We have now found that $2^{\prime}$ hydroxychalcones $\mathbf{1}$ are converted into flavanones $\mathbf{2}$ very efficiently in a water suspension medium and the products isolated simply by filtration. An efficient cyclization reaction of $2^{\prime}$-hydroxychalcones 1 to 2,3-dihydroflavanols 3 by using $\mathrm{NaOH}-\mathrm{H}_{2} \mathrm{O}_{2}$ in a water suspension medium is also reported. These reactions require no organic solvent (except for product recrystallisation), and waste minimization, simple operation, and easier product work-up can be achieved.

## Results and discussion

It has been reported that the intramolecular cyclization reaction of 1a in MeOH using NaOH as base gives flavanone 2a in only $20 \%$ yield at room temperature after 2-3 days. ${ }^{3}$ Very interestingly, however, when the reaction was carried out in a water suspension medium in the presence of surfactants, flavanone 2a was obtained in quantitative yield. For example, a suspension of a mixture of powdered $2^{\prime}$-hydroxychalcone $1 \mathbf{1 a}(1.0 \mathrm{~g}, 4.5$ $\mathrm{mmol}), \mathrm{NaOH}(8 \mathrm{M}, 0.1 \mathrm{ml})$ and sodium 1-dodecane sulfonic acid $(0.01 \mathrm{~g})$ in water $(10 \mathrm{ml})$ was stirred at room temperature for 1 h . The crude product was collected by filtration, washed with water and dried in a desiccator to give flavanone 2a ( 0.95 g, $95 \%$ yield). However, the reaction using no surfactant gave 2a in only $13 \%$ yield. Similarly, tetrabutylammonium iodide, tetrabutylphosphonium bromide, hexadecyltrimethylammonium bromide, glycine, l-alanine, L-proline and L-leucine were also effective for the conversion of $\mathbf{1 a}$ to $\mathbf{2 a}$ in a water suspension medium (Table 1)

It is also found that piperidine catalysed intramolecular cyclization reaction of $\mathbf{1}$ into $\mathbf{2}$ proceeds very efficiently in a water suspension medium (Table 2). The preparation of flavanone 2a is representative of the general procedure employed. For example, a suspension of powdered $2^{\prime}$-hydroxychalcone 1a ( $1.0 \mathrm{~g}, 4.5 \mathrm{mmol}$ ) in water $(10 \mathrm{ml})$ containing piperidine ( $0.01 \mathrm{~g}, 0.12 \mathrm{mmol}$ ) was stirred at room temperature for 1 h . The crude product was collected by filtration, washed with water and dried in a desiccator to give flavanone 2a ( 0.98
$\mathrm{g}, 98 \%$ yield). Similarly, chalcones $\mathbf{1 b}, \mathbf{1 c}, \mathbf{1 e}, \mathbf{1 f}$ and $\mathbf{1 g}$ gave the corresponding flavanones $2 \mathbf{b}, \mathbf{2 c}, \mathbf{2 e}, 2 \mathbf{f}$ and $\mathbf{2 g}$, whilst $\mathbf{1 d}$ yielded a $45: 55$ mixture of $\mathbf{1 d}$ and $\mathbf{2 d}$ (Table 2).
2,3-Dihydroxyflavonol 3a has been reported to be formed in poor yield by treatment of $2^{\prime}$-hydroxychalcone 1a with $\mathrm{NaOH}-$ $\mathrm{H}_{2} \mathrm{O}_{2}$ in organic solvents (Table 3). ${ }^{7,8}$ We have found that the conversion of 1a into 3a proceeds more efficiently in a water suspension medium and the products are isolated simply by filtration. For example, a suspension of a mixture of powdered $2^{\prime}$-hydroxychalcone 1a ( $0.10 \mathrm{~g}, 0.45 \mathrm{mmol}$ ), an aq. NaOH solution ( $8 \mathrm{M}, 1.0 \mathrm{ml}$ ) and a $30 \%$ hydrogen peroxide solution $(0.25 \mathrm{~g})$ was stirred at room temperature for 2 h . The crude

Table 1 NaOH catalysed cyclization of $2^{\prime}$-hydroxychalcone 1a to flavanone 2a


1a
2a

| $\mathbf{1 a}$ | 2a |
| :--- | :--- |
| Additive | Yield (\%) |
| None | 13 |
| $\mathrm{Bu}^{\mathrm{n}} \mathrm{N}^{+} \mathrm{I}^{-}$ | 82 |
| $\mathrm{Bu}^{n}{ }_{4} \mathrm{P}^{+} \mathrm{Br}^{-}$ | 80 |
| $\mathrm{C}_{16} \mathrm{H}_{33} \mathrm{~N}^{+} \mathrm{Me}_{3} \mathrm{Br}^{-}(\mathrm{CTAB})$ | 92 |
| $\mathrm{C}_{12} \mathrm{H}_{25} \mathrm{SO}_{3}-\mathrm{Na}^{+}(\mathrm{SDS})$ | 95 |
| Glycine | 96 |
| L-Alanine | 99 |
| L-Proline | 92 |
| L-Leucine | 93 |

## Green Context

The simplicity of a chemical process often correlates well with its 'greeness'. Multistep or multi-component reactions almost invariably lead to waste if only as a result of demanding separation steps. Here we see some examples of relatively simple one-step reactions that do not use organic solvents either in the reaction or in the work-up. By running the reaction as a suspension in water, the organic product can be separated simply by filtration with no work-up required. Simple basic catalysts are used to ensure fast reactions under moderate conditions. In one case hydrogen peroxide is used as an in-situ oxidant-ideal as the byproduct in the reaction medium, water.

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Table 2 Piperidine catalyzed cyclization reaction of 2'-hydroxychalcones $\mathbf{1}$ to flavanones $\mathbf{2}$ in a water suspension medium

| Compound | Ar | Yield (\%) |
| :--- | :--- | :--- |
| $\mathbf{a}$ | Ph | 98 |
| $\mathbf{b}$ | $o-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | 95 |
| $\mathbf{c}$ | $m-\mathrm{MeOCH}_{6} \mathrm{H}_{4}$ | 96 |
| $\mathbf{d}$ | $p-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | 55 |
| $\mathbf{e}$ | $o-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | 93 |
| $\mathbf{f}$ | $m-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | 94 |
| $\mathbf{g}$ | $p-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | 97 |

Table 3 Conversion of 2'-hydroxychalcone 1a to 2,3-dihydroflavonol 3a


| 1a |  |  | 3a |  |
| :--- | :--- | :--- | :---: | :---: |
| Reagent | Solvent | Temp. | Time/h | Yield (\%) |
| $\mathrm{NaOH}-\mathrm{H}_{2} \mathrm{O}_{2}{ }^{a}$ | MeOH | rt | 3 | 49 |
| $\mathrm{Et}_{2} \mathrm{NH}-\mathrm{H}_{2} \mathrm{O}_{2}{ }^{b}$ | dioxane | $<5^{\circ} \mathrm{C}$ | 40 | 38 |
| $\mathrm{NaOH}-\mathrm{H}_{2} \mathrm{O}_{2}$ | $\mathrm{H}_{2} \mathrm{O}$ | rt | 2 | 100 |
| ${ }^{a}$ Ref. 7. ${ }^{b}$ Ref. 8. |  |  |  |  |

product was collected by filtration, washed with water and dried in a desiccator to give flavanol $\mathbf{3 a}$ ( $0.10 \mathrm{~g}, 100 \%$ yield).

## Experimental

## Typical procedure for the conversion of $2^{\prime}$-hydroxychalcones 1 into flavanones 2 in a water suspension medium

Crystals of 1a were finely powdered by grinding with a pestle and mortar for a few minutes. A suspension of a powdered $2^{\prime}$ hydroxychalcone $\mathbf{1 a}(1.0 \mathrm{~g}, 4.5 \mathrm{mmol})$ was stirred in water (10
$\mathrm{ml})$ containing piperidine $(0.01 \mathrm{~g}, 0.12 \mathrm{mmol})$ at room temperature for 1 h . The crystalline powder formed was filtered off, washed with water and dried in a desiccator to give flavanone 2a ( $0.98 \mathrm{~g}, 98 \%$ yield). The crude crystals thus obtained were recrystallized from EtOH to give pure 2a as colorless needles. Data for 2a; mp 75-76 ${ }^{\circ} \mathrm{C}$; $v(\mathrm{C}=\mathrm{O}) 1718$ $\mathrm{cm}^{-1} ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{Me}_{4} \mathrm{Si}\right) 7.04-7.96(9 \mathrm{H}, \mathrm{m}), 5.50$ ( 1 H , dd, J 3.0, 13.2), 3.11 ( 1 H , dd, J 13.2, 16.8), 2.90 ( 1 H , dd, J 3.0, 16.8).

## Typical procedure for the conversion of $2^{\prime}$-hydroxychalcones 1 into flavanols 3 in a water suspension medium

Crystals of 1a were finely powdered by grinding with a pestle and mortar for a few minutes. A suspension of a mixture of powdered $2^{\prime}$-hydroxychalcone $1 \mathbf{1 a}(0.10 \mathrm{~g}, 0.45 \mathrm{mmol})$, an aq. NaOH solution ( $8 \mathrm{M}, 1.0 \mathrm{ml}$ ) and a $30 \%$ hydrogen peroxide solution $(0.25 \mathrm{~g})$ was stirred at room temperature for 2 h . The crude product was filtered off, washed with water and dried in a desiccator to give flavanol 3a ( 0.10 g , $100 \%$ yield). Recrystallization of the crude product from MeOH gave pure 3a as colorless needles. Data for 3a; $\mathrm{mp} 178-180^{\circ} \mathrm{C} ; v(\mathrm{OH}) 3675$ $(\mathrm{OH}), v(\mathrm{C}=\mathrm{O}) 1718 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{Me}_{4} \mathrm{Si}\right)$ 7.04-7.94 ( $9 \mathrm{H}, \mathrm{m}$ ), 5.14 ( $1 \mathrm{H}, \mathrm{d}, J 12.0$ ), $4.64(1 \mathrm{H}, \mathrm{d}, J 12)$.

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